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SYNTHESIS AND ANTIBACTERIAL ACTIVITY OF 2-ALKOXY PENEMS

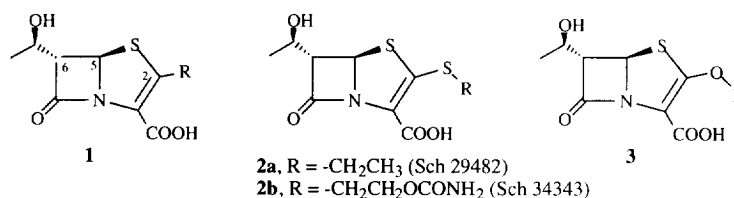
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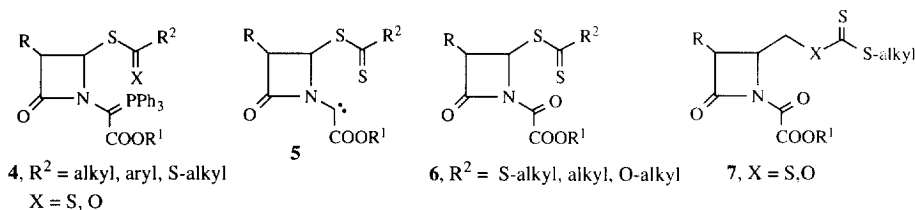
Abstract: The phosphite mediated Oxalimide cyclization reaction was extended to 4-dithiocarbonates of N-oxalyl-2-azetidinones to synthesize 2-alkoxy penems **3**. In general, the *in vitro* antibacterial potency of compounds **3** was weak compared to the highly potent 2-alkylthiopenems **2**.

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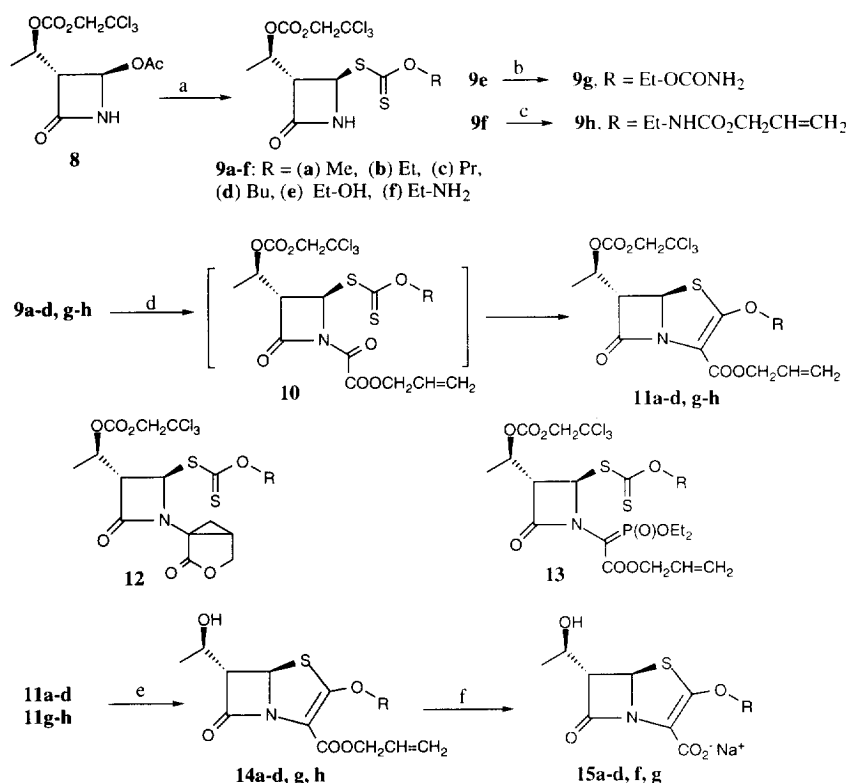
Penems **1** were the subject of intense synthetic studies by several research groups during the 1980s in view of their potential clinical use as potent broad spectrum antibacterial agents. From our laboratories, such investigations led to the identification of the two potent alkylthiopenems **2** as candidates for clinical development as broad spectrum antibacterials.¹ We report here the synthesis and antibacterial activity of 2-alkoxy penems **3**, which include the oxa analogs of **2**.



The Woodward penem synthetic methodology that utilizes the cyclization of phosphorane intermediates **4** has been used extensively in penem research. However, this methodology is not applicable to dithiocarbonates **4** (R² = O-alkyl, X = S) to generate 2-alkoxy penems **3** and consequently prior literature reports have described multistep reaction sequences for synthesizing these compounds.² The oxalimide cyclization reaction, which proceeds via carbene intermediates **5**, generated from oxalimides **6** by trialkylphosphites, is a convenient method to prepare 2-alkylthiopenems³ and 2-alkylpenems^{4a} from appropriate intermediates **6**, and 3-heterosubstituted isocephems and iso-oxacephems from intermediates **7**.^{4b} We have found that the oxalimide reaction is also a simple and convenient method to synthesize the 2-oxapenems **3** described herein (Scheme 1).⁵



Thus, displacement of the C₄-acetoxy of **8**⁶ with various xanthates (ROCSS[−]Na⁺) derived from the corresponding alcohols provided the C₄-dithiocarbonates **9a–f** (58–66%). Dithiocarbonates **9e–f** were functionalized (**9e**, OH to OCONH₂) or protected (**9f**, NH₂ to NH-CO₂CH₂CH=CH₂) to afford **9g–h**. The dithiocarbonates **9a–d** and **g–h** were converted (acylation with allyloxalyl chloride) to the oxalimides **10**.

Scheme 1^a

^aReagents and conditions: (a) ROH/1 N NaOH (1 equiv.)/CS₂/THF/rt. (b) NaCNO/CF₃COOH (1.1 equiv.)/CH₂Cl₂/rt. (c) ClC(O)OCH₂CH=CH₂/Et₃N/CH₂Cl₂. (d) i. ClC(O)CO₂CH₂CH=CH₂/i-Pr₃NEt/CaCO₃/CH₂Cl₂/0 °C. ii. CHCl₃/P(OEt)₃ (1 equiv.)/syringe pump addition/reflux 17 h/chromat. (e) Zn dust/THF-50% AcOH-10% H₂O/-15 °C/chromat. (f) for **14a–d, g**: Sodium 2-Et-hexanoate (1 equiv.)/Pd(PPh₃)₄ (cat)/PPh₃/EtOAc, rt; for **14h**: pyridinium formate/Pd(PPh₃)₄ (cat)/PPh₃/CH₂Cl₂, rt.

which without purification were treated with triethylphosphite (slow addition) in refluxing chloroform to afford after chromatography the desired protected 2-alkoxypenems **11a–d** and **11g–h** (30–40%). The expected competing products of the oxalimide cyclization,³ cyclopropanes **12** and phosphoranes **13**, were formed in low yields in all of these cyclization reactions. Removal of the trichloroetoxycarbonyl protective groups (zinc dust/dil. acetic acid) afforded **14a–d, g, h**. Allylester exchange⁷ mediated by Pd⁰ for **14a–d, g** provided the 2-alkoxypenems **15a–d, g** as their sodium salts; palladium⁰ mediated allyl deprotection under reductive conditions⁸ was used for the deprotection of **14f** to **15f**.¹⁰

Table 1. GRAM-POSITIVE *IN VITRO* ANTIBACTERIAL ACTIVITY OF 2-ALKOXY PENEMS

Entry Number	Geometric Mean MICs in µg/mL (number of strains)			GRAM-POSITIVE Geometric Mean MICs µg/mL
	<i>B. subtilis</i>	<i>Staphyl.</i>	<i>E. faecalis</i>	
15a	1.41 (2)	1.51 (21)	32.0 (4)	3.20 (28)
15b	1.0 (2)	3.12 (28)	64.0 (2)	2.80 (31)
15c	1.0 (1)	1.66 (22)	32.0 (4)	2.44 (28)
15d	1.0 (1)	1.51 (22)	32.0 (4)	2.26 (28)
15f	8.0 (1)	27.43 (18)	45.26 (2)	27.34 (22)
15g	1.0 (1)	1.57 (17)	64.0 (2)	2.07 (21)
2a				0.10 **
2b				0.20 **
2c				0.20 **

* Mueller–Hinton agar, 24 h; ** Data from ref. 1

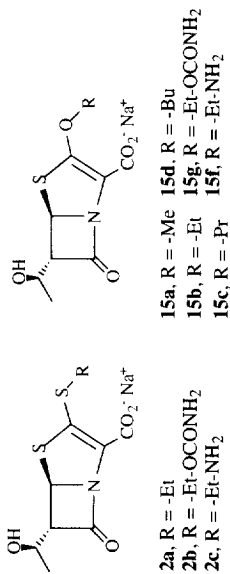


Table 2. GRAM-NEGATIVE *IN VITRO* ANTIBACTERIAL ACTIVITY OF 2-ALKOXY PENEMS*

Entry Number	Geometric Mean MICs in µg/mL (number of strains)								GRAM-NEGATIVE Geometric Mean MICs µg/mL	
	<i>E. coli</i>	<i>Enterobacter</i>	<i>Klebsiella</i>	<i>Morganella</i>	<i>Providencia</i>	<i>Salmonella</i>	<i>Sarcinea lut.</i>	<i>Serratia</i>	Geometric Mean	MICs
15a	10.44 (13)	24.25 (5)	8.98 (12)	45.26 (2)	39.40 (10)	6.96 (5)	2.0 (1)	29.63 (9)	16.60	(56)
15b	6.82 (13)	≥24.25 (5)	5.04 (12)	45.26 (2)	39.40 (10)	4.60 (5)	1.0 (1)	≥22.63 (8)	12.13	(55)
15c	14.38 (13)	≥73.52 (5)	20.16 (12)	64.0 (2)	≥64.0 (9)	10.56 (5)	1.0 (1)	≥87.09 (9)	31.60	(55)
15d	≥25.85 (13)	≥128.0 (5)	≥32 (12)	≥90.51 (2)	≥80.64 (9)	≥36.76 (5)	1.0 (1)	≥128 (9)	52.98	(55)
15f	≥128.0 (11)	≥128.0 (3)	≥128.0 (10)	≥128.0 (1)	≥128.0 (5)	≥128.0 (3)	32.0 (1)	≥128 (4)	≥128.0	(37)
15g	6.62 (11)	10.08 (3)	4.92 (10)	16.0 (1)	27.86 (5)	4.0 (3)	0.50 (1)	16.0 (4)	8.30	(37)
2a									0.70	**
2b									0.50	**
2c									3.60	**

* Mueller–Hinton agar, 24 h; ** Data from ref. 1

The *in vitro* antibacterial activity of the 2-alkoxy penems is summarized in Tables 1 and 2. Methodology for determining the minimum inhibitory concentrations (MIC values) has been described previously.⁹ In general, the 2-alkoxy penems have modest activity against gram-positive organisms and weak activity against gram-negative organisms. The gram-negative activity decreases when the alkyl group is larger than ethyl (e.g., **15b** to **15c** and **15d**). An amino substituent on the ethyl group has a deleterious effect on the gram-negative activity (**15f** vs. **15b**). When compared to the C₂-alkylthio series (**2a-c**), the 2-alkoxy penems are approximately 10x less active against gram-positive organisms and 20x less active against gram-negative organisms.

In summary, the oxalimide cyclization reaction has been applied for the synthesis of a selected series of 2-alkoxy penems. The compounds were found to be weak antibacterial agents when compared to the corresponding 2-alkylthio penems, thereby demonstrating the importance of C₂-sulfur atom for antibacterial potency in penems.

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10. All new compounds gave mass spectral and spectroscopic data in agreement with their assigned structures. Data on selected compounds: **9c**: ¹H NMR (CDCl₃, 80 MHz) δ 1.02 (t, 3H, *J* = 6 Hz), 1.50 (d, 3H, *J* = 6 Hz), 1.79 (q, 2H, *J* = 6 Hz), 3.40 (dd, 1H, *J* = 2, 4 Hz), 4.58 (t, 2H, *J* = 6 Hz), 4.80 (s, 2H), 5.27 (m, 1H), 5.50 (d, 1H, *J* = 2 Hz). **11b**: white crystals from EtOAc/hexane, mp 84–85 °C; [α]_D²⁰ +167° (CHCl₃, c 0.3); λ_{max} (MeOH) 310 mμ (ε 9000), 275 mμ (ε 6000), 206 mμ (ε 7400); IR (CHCl₃) 1798, 1760 cm⁻¹; ¹H NMR (CDCl₃, 80 MHz) δ 1.44 (t, 3H, *J* = 6 Hz), 1.54 (t, 3H, *J* = 6 Hz), 3.87 (dd, 1H, *J* = 1.5, 9 Hz), 4.22 (q, 2H), 4.70 (m, 3H), 4.77 (s, 2H), 5.25 (m, 2H), 5.60 (d, 1H, *J* = 1.5 Hz), 5.90 (m, 1H). **14c**: ¹H NMR (CDCl₃, 80 MHz) δ 0.98 (t, 3H, *J* = 7 Hz), 1.34 (d, 3H, *J* = 6 Hz), 3.64 (dd, 1H, *J* = 1.7, 7 Hz), 4.1 (m, 3H), 4.67 (m, 2H), 5.30 (m, 2H), 5.57 (d, 1H, *J* = 1.7 Hz), 5.93 (m, 1H). **15d**: ¹H NMR (D₂O, 80 MHz) δ 0.90 (t, 3H, *J* = 6 Hz), 1.30 (d, 2H, *J* = 6 Hz), 3.79 (dd, 1H, *J* = 1.5, 6 Hz), 4.12 (m, 3H), 5.57 (d, 1H, *J* = 1.5 Hz). **15g**: ¹H NMR (D₂O, 80 MHz) δ 3.90 (dd, 1H, *J* = 1.5, 6 Hz), 5.67 (d, 1H, *J* = 1.5 Hz).